

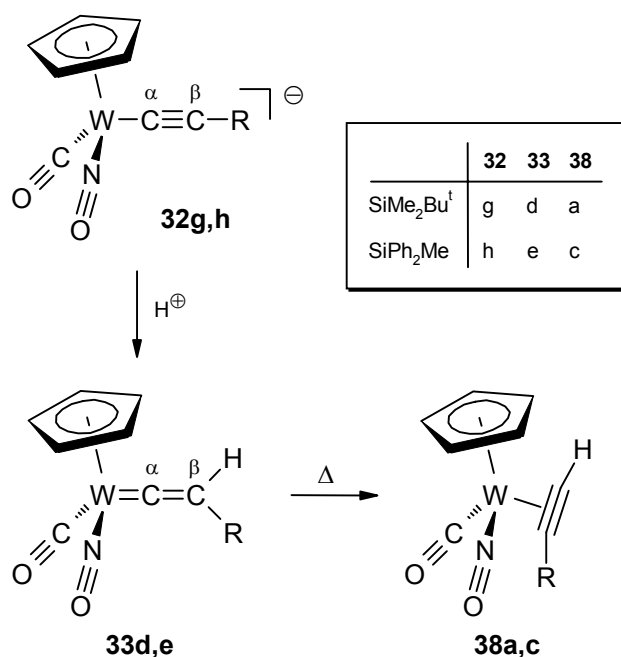
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## Mechanistic Studies of the $\eta^1$ -Vinylidene $\rightarrow$ $\eta^2$ -Alkyne Isomerization and Synthetic Application of Carbonyl-Vinylidene Complexes of Tungsten

### Abstract

#### Part 1: $\eta^1$ -Vinylidene $\rightarrow$ $\eta^2$ -Alkyne Isomerization

In the first part the synthesis, characterization and isomerization of tungsten vinylidene complexes containing a silyl group at the  $\beta$ -carbon atom is described. Protonation of lithium tungsten acetylides  $[W(-C\equiv C-R)Cp(CO)(NO)]Li$  [ $R = SiMe_2Bu^t$  (**32g**),  $R = SiPh_2Me$  (**32h**)] lead to the formation of the corresponding vinylidene complexes  $[W(=C=CHR)Cp(CO)(NO)]$  (**33d,e**). It is shown that contrary to the general prediction, these monosubstituted vinylidenes convert thermally into the corresponding  $\eta^2$ -alkyne complexes  $[W(\eta^2-H-C\equiv C-R)Cp(CO)(NO)]$  (**38a,c**). Kinetic studies on the transformation of **33d** to **38a** provided following activation parameters:  $\Delta H^\ddagger = 27.2 \pm 0.5$  kcal mol $^{-1}$  and  $\Delta S^\ddagger = -9.7 \pm 1.1$  eu. These values, the measured kinetic isotope effect of **33d**  $\rightarrow$  **38a** and the synthetic data are consistent with the isomerization mechanism via a sigmatropic 1,2-shift of the silyl group.



#### Part 2: Nucleophilic Addition to Carbonyl-Vinylidene Complexes

Primary and secondary amines have been found to react in THF at room temperature with vinylidene complexes  $[W(=C=CHR)Cp(CO)(NO)]$  [ $R = C(CH_3)_3$  (**33a**),  $R = SiMe_2Bu^t$  (**33d**)] affording aminocarbene derivatives  $[W\{=C(NR^1R^2)CH_2C(CH_3)_3\}Cp(CO)(NO)]$  and  $[W\{=C(NR^1R^2)CH_2SiMe_2Bu^t\}Cp(CO)(NO)]$  [ $R^1 = H$ ,  $R^2 = n-C_4H_9$  (**90a**, **91a**);  $R^1 = H$ ,  $R^2 = CH(CH_3)_2$  (**90b**, **91b**);  $R^1 = H$ ,  $R^2 = C(CH_3)_3$  (**90c**, **91c**);  $R^1 = R^2 = (CH_2)_4$  (**90d**, **91d**)]. At short reaction time the nucleophilic addition of both primary and secondary amines to **33a** occurs at the carbonyl carbon atom leading stereoselectively to the thermodynamically less stable  $\eta^2$ -carbamoyl-(*Z*)-vinyl complexes  $[W\{\sigma-(Z)-CH=CHC(CH_3)_3\}\{\eta^2-C(O)NR^1R^2\}Cp(NO)]$  [ $R^1 = H$ ,  $R^2 = n-C_4H_9$  (**92a**);  $R^1 = H$ ,  $R^2 = CH(CH_3)_2$  (**92b**);  $R^1 = H$ ,  $R^2 = C(CH_3)_3$  (**92c**);  $R^1 = R^2 = (CH_2)_4$  (**92d**)]. At higher amine concentration and prolonged reaction time **92a-d** form the

corresponding aminocarbene complexes **90a-d**. The structure of **92b** has been authenticated by a single X-ray diffraction analysis.

The outcome of the additions depends on the vinylidene precursor leading stereoselectively to  $\eta^2$ -carbamoyl-(Z)-vinyl complexes **92a-d** in a short reaction time when a *tert*-butyl substituted vinylidene complex is used as a precursor. In the presence of an excess of the appropriate amine the complexes **92a-d** are not stable and undergo an elimination reaction to generate the starting materials. With a longer reaction time the amine attacks again the complex **33a** and leads via addition to the C $_{\alpha}$  of the vinylidene moiety to the aminocarbenes **90a-d**. These results show a novel reactivity pattern in the nucleophilic addition of amines to vinylidene complexes bearing carbonyl ligands as competitive electrophilic sites.

